



**PROGRAMS AND BIBLIOGRAPHY**

<b>Subject</b>	
<b>Code</b>	<b>Name</b>
QI245	Solid State Chemistry

<b>Vector</b>
OF:S-5 T:002 P:000 L:000 O:000 D:000 HS:002 SL:002 C:002 AV:N EX:S FM:75%

<b>Pre requirement</b>
QI145

<b>Summary</b>
Close packed structures. Some important crystalline structure types. X-ray diffraction. Defects and non-stoichiometric compounds. Electronic, optical and magnetic properties of solids.

<b>Program</b>
Close packed structures. Unit cells, crystal system and Bravais lattice. Principles of X-Ray diffraction. Lattice planes and Miller indices. Crystallographic card. Some important crystalline structure types (CsCl, NaCl, ZnS, CaF <sub>2</sub> , among others).  Defects in ionic crystals. Stoichiometric defects: intrinsic defects (Schottky and Frenkel) and extrinsic defects (solid solution). Non-stoichiometric. Ionic conductivity.  Electronic conductivity in solids: molecular orbital theory and energy-band model (metals, semiconductors, and insulators). Intrinsic and extrinsic semiconductors. Electronic conductivity as a function of temperature.  Optical properties: ruby laser, neodymium laser and light-emitting diodes.  Magnetic properties: magnetic susceptibility, magnetism in metals. Ferromagnetism, ferrimagnetism and antiferromagnetism.

## **Bibliography**

### **Textbooks**

- L. E. Smart, E. A. Moore. Solid State Chemistry: An Introduction. Boca Raton: CRC, 2012. 465p.  
A. R. West. Basic Solid-State Chemistry. 2<sup>nd</sup> ed. Chichester: John Wiley, 1999. 480p.  
W.D. Callister. Ciência e Engenharia de Materiais: uma Introdução, 8<sup>a</sup>. ed. Rio de Janeiro: LTC, 2012. 817p.  
C. E. Housecroft, A. G. Sharpe. Inorganic Chemistry. 4th ed. Upper Saddle River. NJ : Prentice-Hall, 2012. 754p.

### **Supplemental Readings**

- D. F. Shriver, P. W. Atkins, C.H. Langford. Inorganic Chemistry. 2<sup>nd</sup>. ed. Oxford: Oxford University Press, 1994. 819p.

## **Evaluation criteria**

For grading policy, see: Regimento Geral de Graduação, Seção I – Normas Gerais, Capítulo V – Da Avaliação do Aluno na Disciplina. Students are required to attend 75 % of the lectures. For further details, see: Regimento Geral de Graduação, capítulo VI, seção X, artigo 72.